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## Key indicators

Single-crystal X-ray study
$T=105 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
$R$ factor $=0.029$
$w R$ factor $=0.074$
Data-to-parameter ratio $=8.7$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## l-Threonyl-t-alanine

In the title compound, $\mathrm{C}_{7} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4}$, peptide molecules are connected into $\beta$-sheet-like ribbons along the short tetragonal axis $\left[4.7065(4) \AA\right.$ ] by one $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and two $\mathrm{C}^{\alpha} ;-\mathrm{H} \cdots \mathrm{O}$ interactions.

## Comment

As part of a systematic survey of dipeptides with one hydrophobic and one hydrophilic residue (Netland et al., 2004), the structure of L-threonyl-L-alanine, TA, has been determined. The molecular structure, with a rather extended peptide main chain (Table 1) is depicted in Fig. 1.


Like l-threonyl-L-valine (TV) (Görbitz, 2005), which was investigated in parallel with TA, the TA crystallization experiments led to the formation of extremely thin needles. While the TV crystals share the hexagonal symmetry of numerous hydrophobic dipeptides (Görbitz, 2003, and references therein), the crystals of TA proved to be tetragonal, with space group $\mathrm{P4}_{2}$. The crystal packing, shown in Fig. 2, is characterized by segregation of hydrophobic groups into two independent types of columns along the tetragonal axis. There are no regular peptides among only $31 P 4_{2}$ entries in the


Figure 1
The molecular structure of TA. Displacement ellipsoids for the non-H atoms are shown at the $50 \%$ probability level. The H atoms are shown as spheres of arbitrary size.

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Figure 2
Molecular packing and unit-cell contents of TA viewed along the $c$ axis. Hydrogen bonding is indicated by dashed lines.


Figure 3
Detail of the hydrogen-bonding network, showing ribbons along the tetragonal axis with a $4.7 \AA$ A repeat unit. For clarity, side chains have been curtailed beyond $\mathrm{C} \beta$.

Cambridge Structural Database (CSD; Version 5.26 of November 2004; Allen, 2002), but related structures in $I 4$ with $Z=8, a \sim 16.0 \AA$ and $c \sim 5.2 \AA$ were observed for L-alanyl-Lalanine (Fletterick et al., 1971) and L-2-aminobutyryl-Lalanine (Görbitz, 2002). The retroanalogue of TA, L-alanyl-Lthreonine (AT) (Netland et al., 2004), has a completely different crystal packing pattern, in space group $P 2_{1} 2_{1} 2_{1}$, but nevertheless incorporates the same $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds, except that the amide H atom is accepted by the threonine side-chain -OH in AT, but by the carbonyl group of the peptide bond in TA (Fig. 2 and Table 2). Together with two additional $\mathrm{C}^{\alpha}-\mathrm{H} \cdots \mathrm{O}$ contacts, the latter interaction forms hydrogen-bonded ribbons parallel to the tetragonal axis, as
shown in Fig. 3, a strong $\beta$-sheet-like synthon that occurs in 13 other dipeptide structures. The 4.7065 (4) $\AA c$ axis of TA represents a new lower limit for the repeat distance of this pattern; the previous range was 4.77-5.26 A. The synthon was last seen in the structure of L-valyl-L-serine (VS) $\left[P 2_{1} ; a=\right.$ 4.7695 (2) Á; Johansen et al., 2005], and it is interesting to note that the hydrogen-bonding pattern of TA in fact can be obtained directly from the rather unusual hydrogen-bonded sheets of VS by folding around the tetragonal axis.

The TA threonyl side-chain -OH also accepts an H atom, but with a methyl donor (Fig. 2 and Table 2). The $-\mathrm{CH}_{3} \cdots \mathrm{OH}$ - type of weak hydrogen bond has not received much attention in the literature, but a search of the CSD demonstrates a fairly high abundance in the crystal structures of organic compounds.

## Experimental

L-Threonyl-L-alanine was obtained from Bachem. Needle-shaped crystals were grown by slow diffusion of acetonitrile into an aqueous solution ( $40 \mu \mathrm{l}$ ) containing about 1 mg of the peptide.

## Crystal data

$\mathrm{C}_{7} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{4}$
$M_{r}=190.20$
Tetragonal, $P 4_{2}$ $a=13.9136$ (7) ${ }^{2} \AA$
$c=4.7065$ (4) $\AA$
$V=911.12(10) \AA^{3}$
$Z=4$
$D_{x}=1.387 \mathrm{Mg} \mathrm{m}^{-3}$

## Data collection

Siemens SMART CCD
diffractometer
$\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.905, T_{\text {max }}=0.994$
6059 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.029$
$w R\left(F^{2}\right)=0.074$
$S=1.09$
1207 reflections
138 parameters
H atoms treated by a mixture of independent and constrained refinement

Table 1
Selected torsion angles $\left({ }^{\circ}\right)$.

| $\mathrm{N} 1-\mathrm{C} 1-\mathrm{C} 4-\mathrm{N} 2$ | $130.52(16)$ | $\mathrm{N} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{O} 2$ | $-59.93(19)$ |
| :--- | :---: | :---: | :---: |
| $\mathrm{C} 1-\mathrm{C} 4-\mathrm{N} 2-\mathrm{C} 5$ | $177.03(15)$ | $\mathrm{N} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | $-178.09(16)$ |
| $\mathrm{C} 4-\mathrm{N} 2-\mathrm{C} 5-\mathrm{C} 7$ | $-137.75(17)$ | $\mathrm{C} 1-\mathrm{C} 2-\mathrm{O} 2-\mathrm{H} 5$ | $87.2(18)$ |
| $\mathrm{N} 2-\mathrm{C} 5-\mathrm{C} 7-\mathrm{O} 3$ | $-16.2(2)$ |  |  |

Table 2
Hydrogen-bond geometry ( $\AA{ }^{\circ},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N} 1-\mathrm{H} 1 \cdots \mathrm{O} 4^{\text {i }}$ | 0.91 (2) | 1.94 (3) | 2.798 (2) | 156 (2) |
| $\mathrm{N} 1-\mathrm{H} 2 \cdots \mathrm{O} 3^{\text {ii }}$ | 0.93 (2) | 2.01 (2) | 2.8822 (19) | 156 (2) |
| $\mathrm{N} 1-\mathrm{H} 3 \cdots \mathrm{O} 4^{\text {iii }}$ | 0.89 (3) | 1.97 (3) | 2.820 (2) | 159 (2) |
| $\mathrm{N} 2-\mathrm{H} 4 \cdots \mathrm{O} 1^{\text {iv }}$ | 0.82 (2) | 2.16 (2) | 2.928 (2) | 156 (2) |
| $\mathrm{O} 2-\mathrm{H} 5 \cdots \mathrm{O} 3^{\text {ii }}$ | 0.84 (3) | 1.90 (3) | 2.7163 (19) | 163 (2) |
| $\mathrm{C} 1-\mathrm{H} 11 \cdots \mathrm{O} 1^{\text {iv }}$ | 0.94 (3) | 2.23 (2) | 3.111 (2) | 155 (2) |
| $\mathrm{C} 3-\mathrm{H} 33 \cdots{ }^{\text {v }}$ | 0.98 | 2.45 | 3.402 (3) | 163 |
| C5-H51 ${ }^{\text {O }} \mathrm{O}^{\text {vi }}$ | 1.00 | 2.50 | 3.405 (2) | 151 |

Symmetry codes: (i) $-y+1, x+1, z-\frac{1}{2}$; (ii) $y,-x+1, z+\frac{1}{2}$; (iii) $-y+1, x+1, z+\frac{1}{2}$; (iv) $x, y, z-1$; (v) $-y+1, x, z+\frac{1}{2}$; (vi) $x, y, z+1$.

Positional parameters were refined for H atoms involved in short hydrogen bonds. Other H atoms were positioned with idealized geometry and fixed $\mathrm{C}-\mathrm{H}$ distances $(0.98-1.00 \AA)$. $U_{\text {iso }}$ values were $1.2 U_{\text {eq }}$ of the carrier atom, or $1.5 U_{\text {eq }}$ for amino and methyl groups. In the absence of significant anomalous scattering effects, 918 Friedel pairs were merged. The absolute configuration was known for the purchased material.

Data collection: SMART (Bruker, 1998); cell refinement: SAINTPlus (Bruker, 2001); data reduction: SAINT-Plus; program(s) used to
solve structure: $\operatorname{SHELXTL}$ (Bruker, 2000); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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